

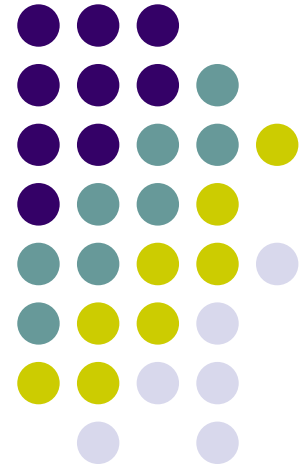
New Directions in Reactor Design Through Miniaturization

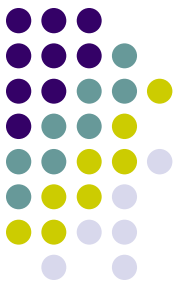
Ronald S. Besser

Professor

Department of Chemical, Biochemical, and Materials
Engineering

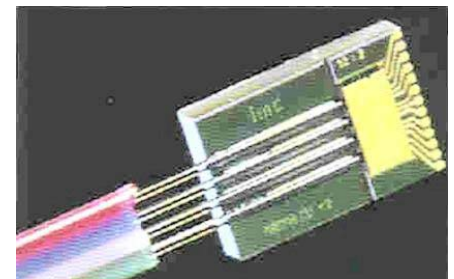
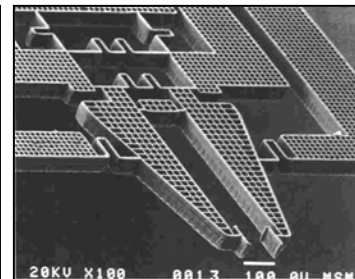
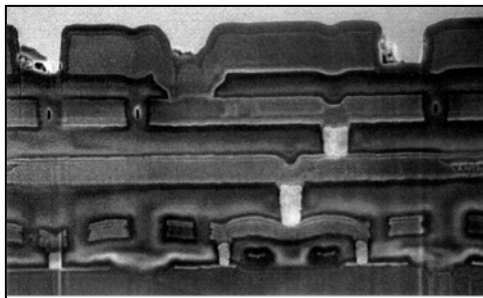
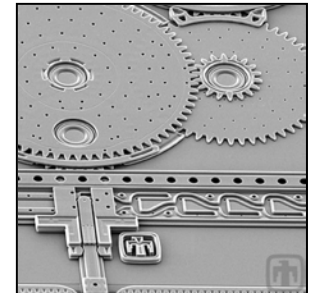
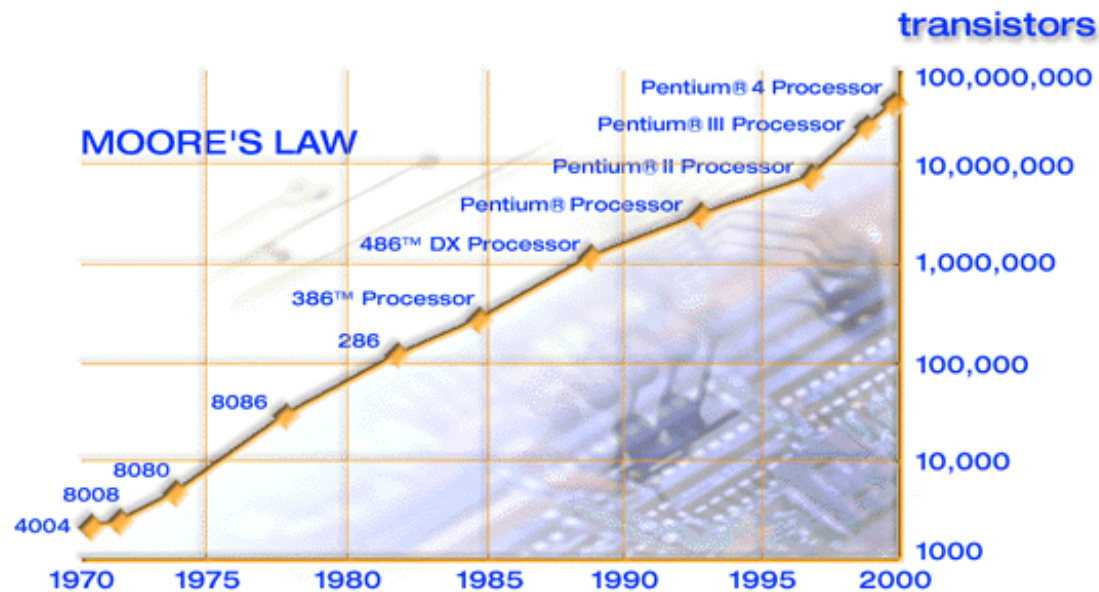
Stevens Institute of Technology
Hoboken, New Jersey



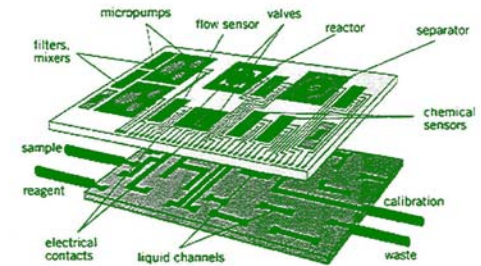
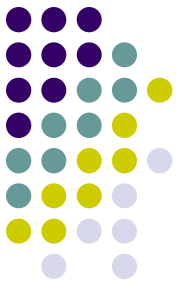


Miniaturization Progress

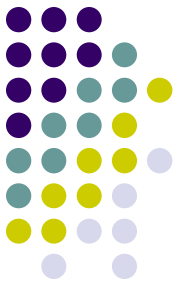
Microelectronics to MEMS



Microchemical Systems



Miniature reaction and other unit operations, possessing ***specific advantages*** over conventional chemical systems



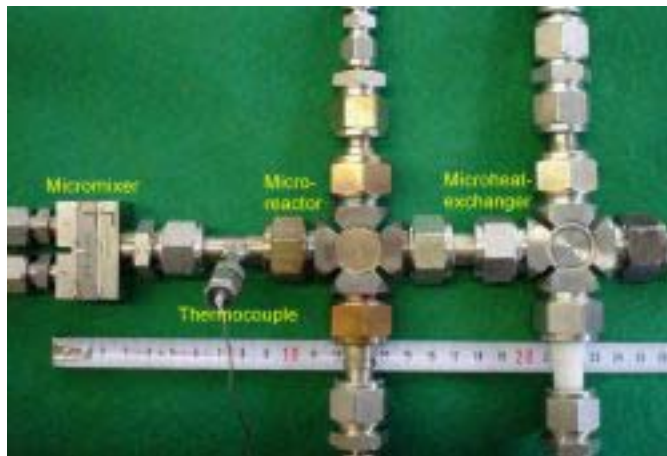
Outline

- Microreactors—What?
- Benefits of Miniaturization—Why?
- Examples of Microreactors: Fabrication, Characterization, and Reaction—How?
- Conclusions

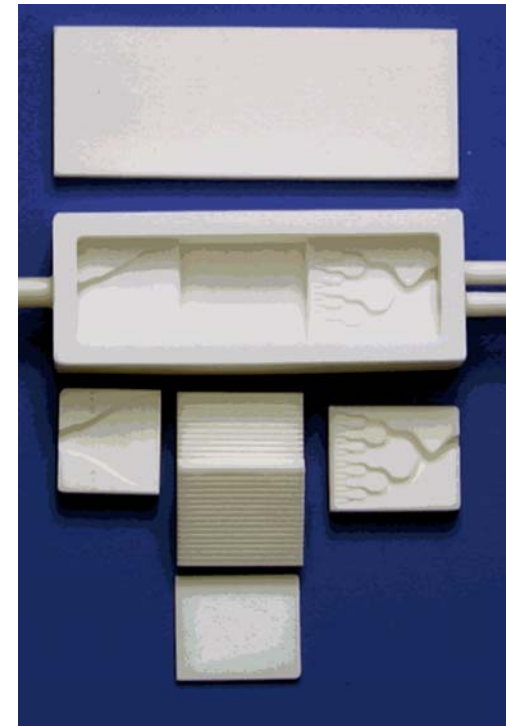
Microreactors—What Are They?



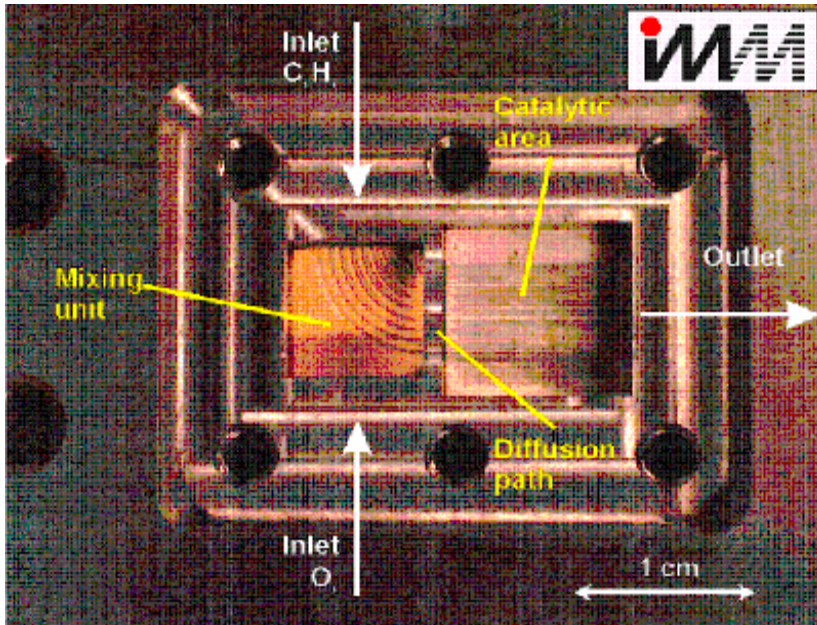
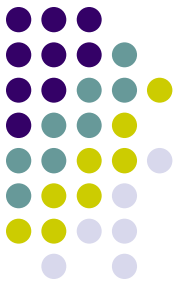
- Not your mother's microreactor
- At least 10X smaller than benchtop “microreactor” of the past



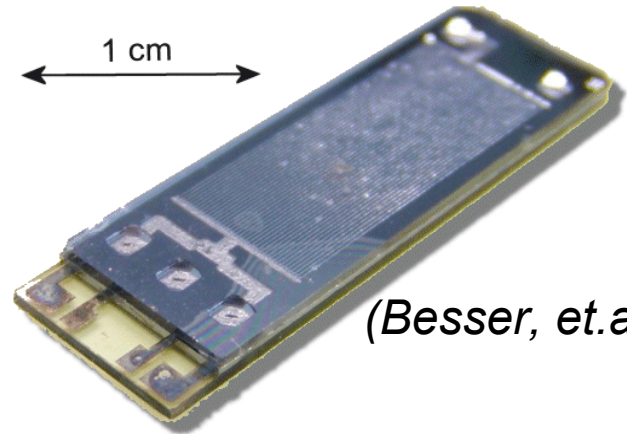
(Forschungszentrum **Karlsruhe** GmbH)



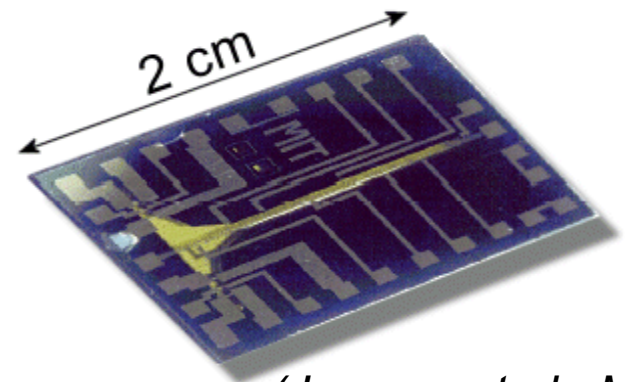
Microreactors—What Are They?



(Ehrfeld, et.al., IMM)

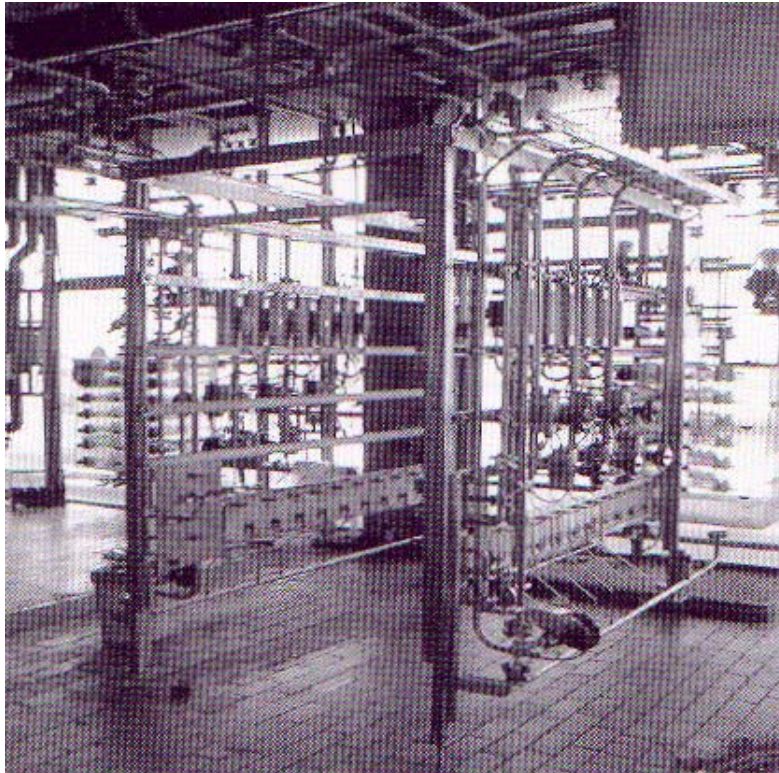
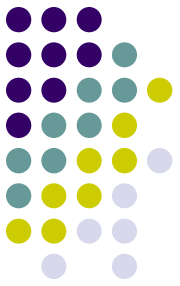


(Besser, et.al., IfM)



(Jensen, et.al., MIT)

Integration-Industrial Processes

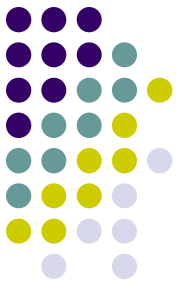


Input flows must be divided and reduced

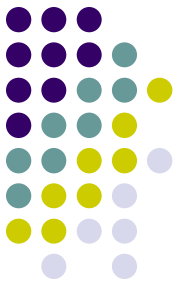
Output flows must be combined

(Merck production plant using micromixers)

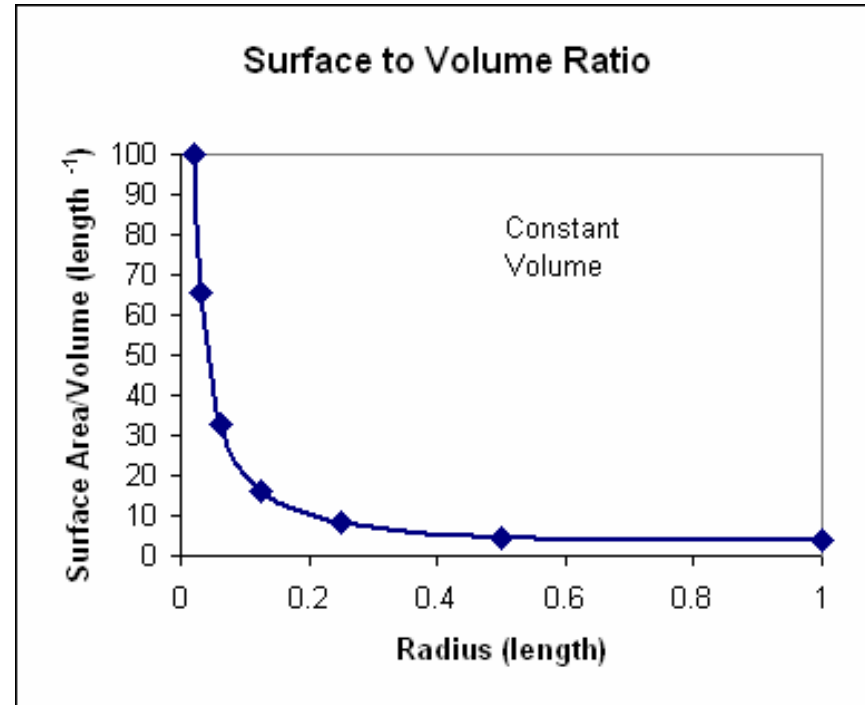
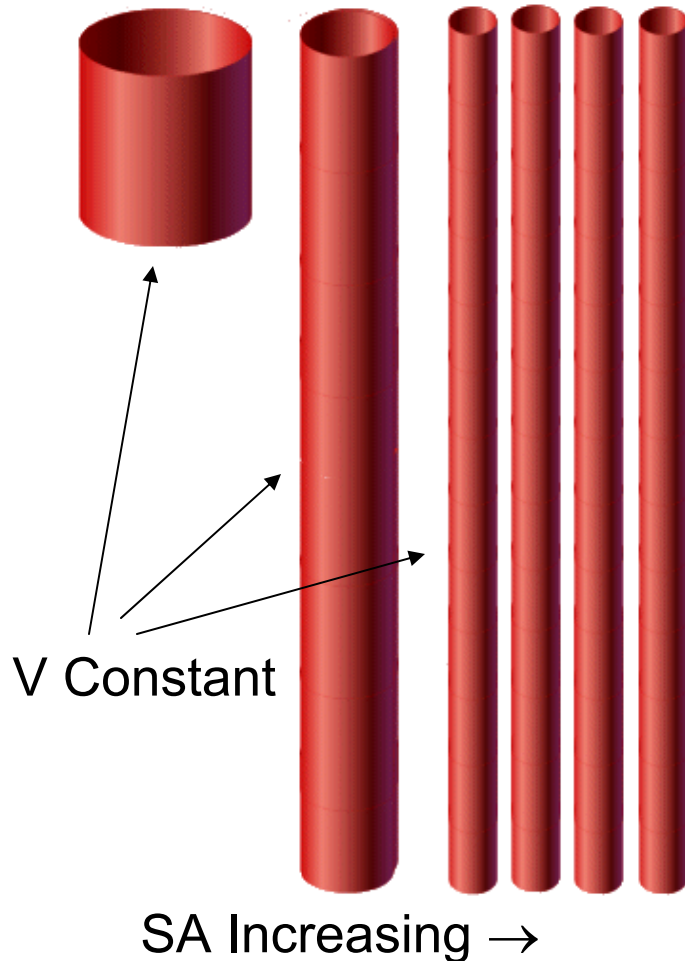
Benefits of Miniaturization— Why?



- Surface to Volume Ratio
- Low Inventory (“Hold Up”)
- Residence Time Distribution
- Low Transport Resistances
- Robust Materials
- Cost

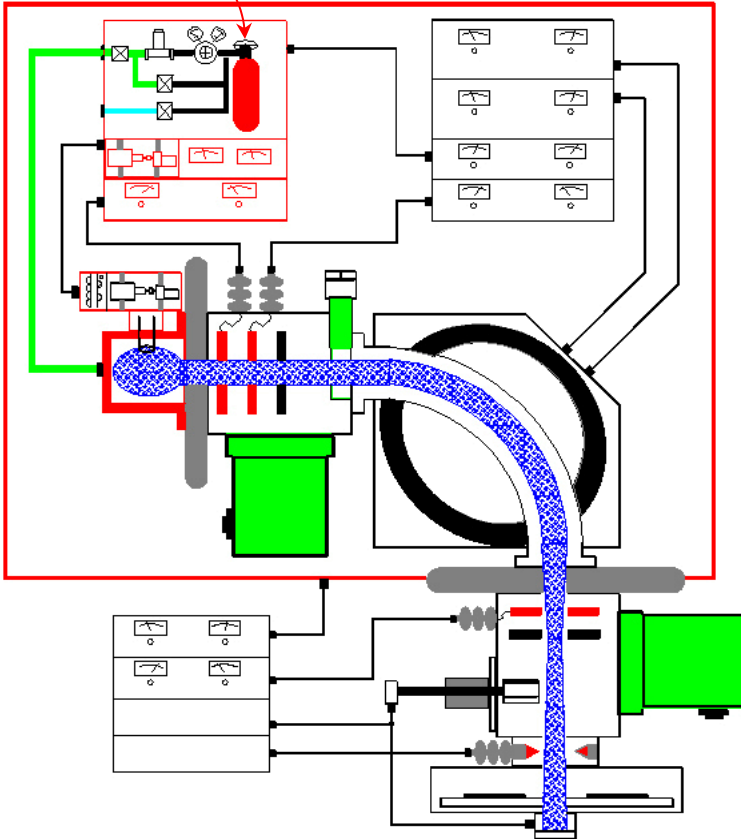


Benefits: Surface to Volume



Heat Management
Surface Reaction
Explosion-Safe

Benefits: Low-Inventory (Hold-Up)



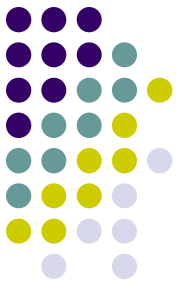
Schematic of As^+ Ion Implanter



- Safety
- Environment

Phosgene Reactor, Geismar, LA

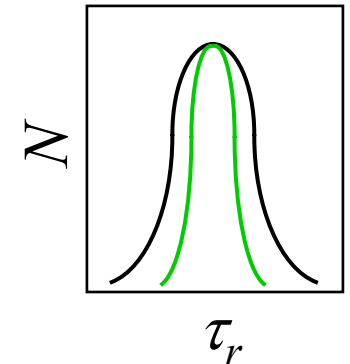
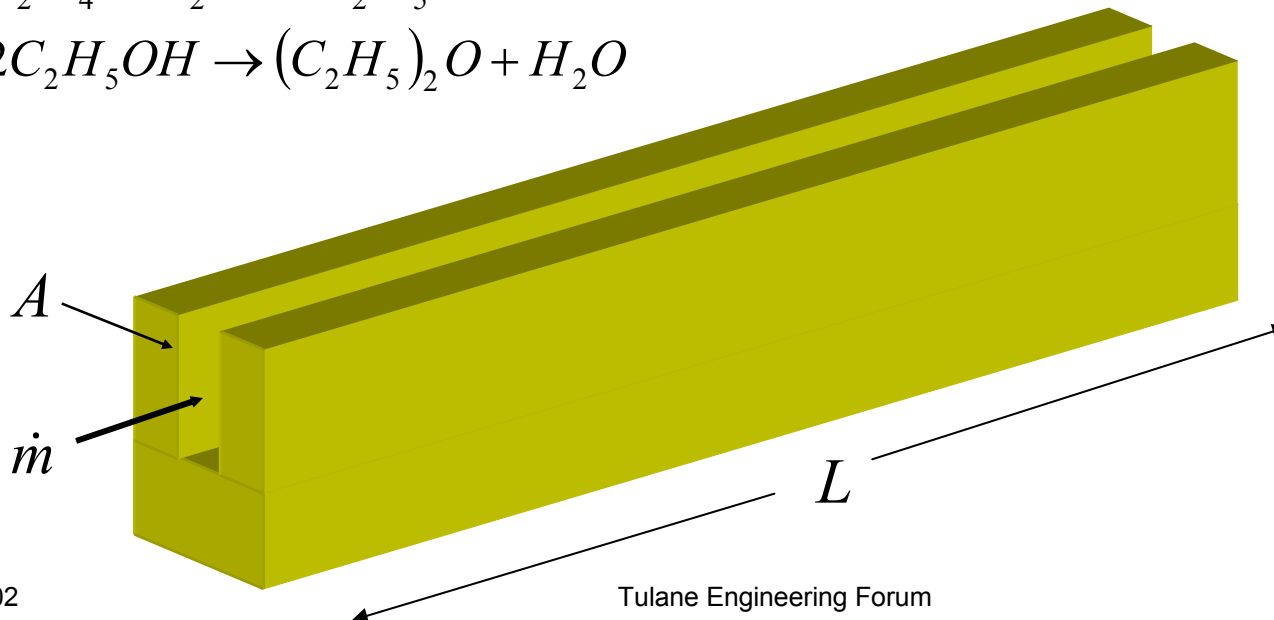
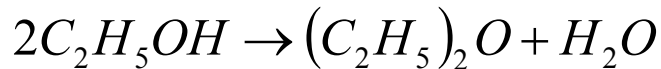
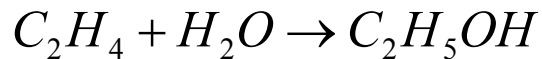
Benefits: Residence Time Distribution



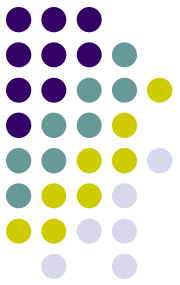
Precise Control Over Geometry

Tuning of residence time

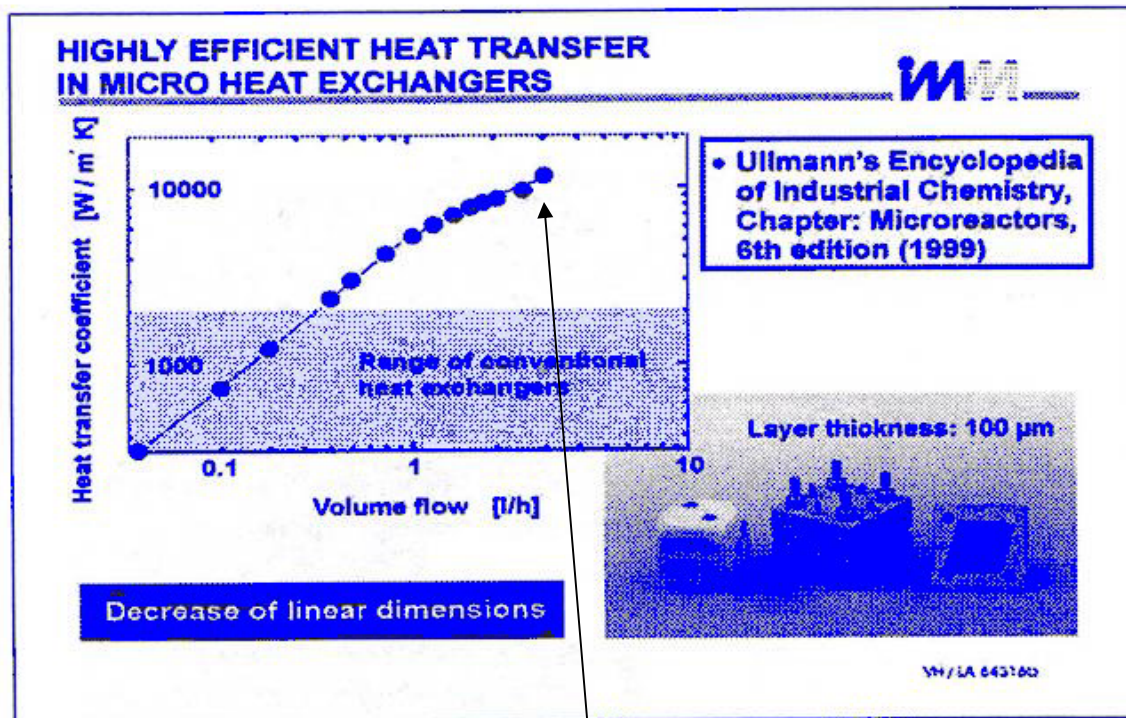
Improved selectivity



Benefits: Low Transport Resistances

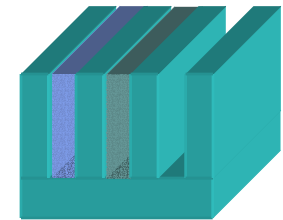


Overall Heat Transfer Coefficient



(AIChE)

$$U = 25,000 \text{ W/m}^2\text{K}$$

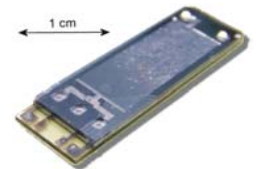
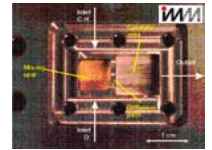
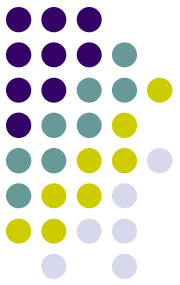


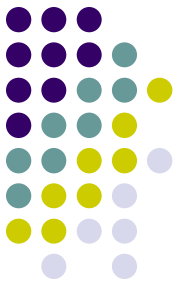
$$q_x'' = -k \frac{dT}{dx}$$

(conduction)

Benefits: Robust Materials

- High strength, high melting point materials:
 - Metals
 - Ceramics
 - Silicon
- Array of fabrication processes (MEMS technology)
- Non-traditional reactor materials
 - Polymers





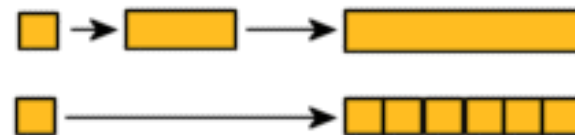
Benefits: Cost

- Reactor Fabrication

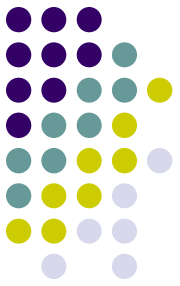
- High volume batch
 - Si integrated circuit fabrication model
 - Metal/ceramic micromachining techniques
 - Interface of reactor to plant (?)

- Scale-Up Process

- Linear process
- Characterize unit module; scale up throughput by addition of modules

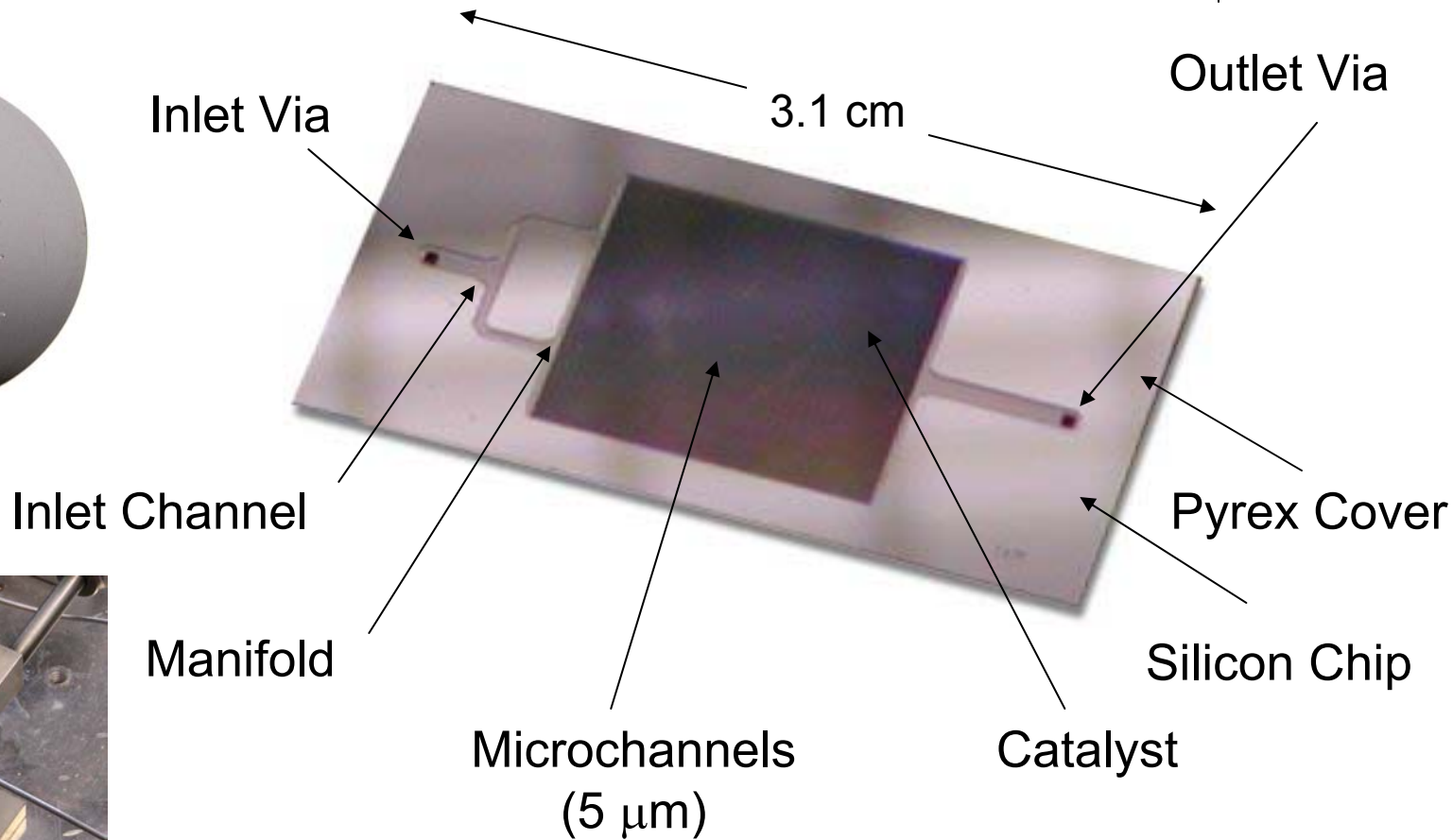
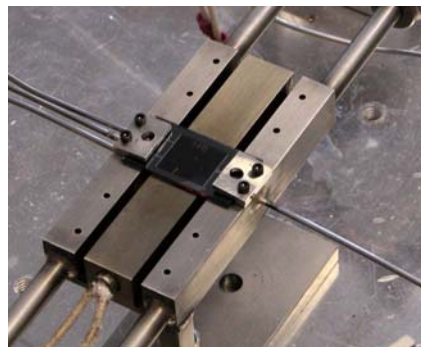
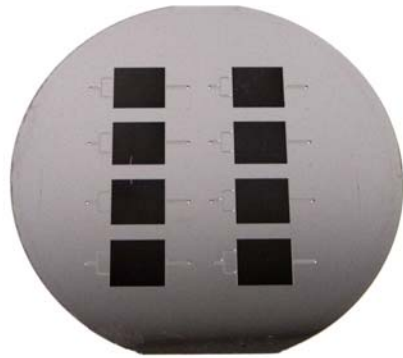
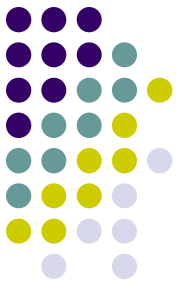


Microreactor Example—How?

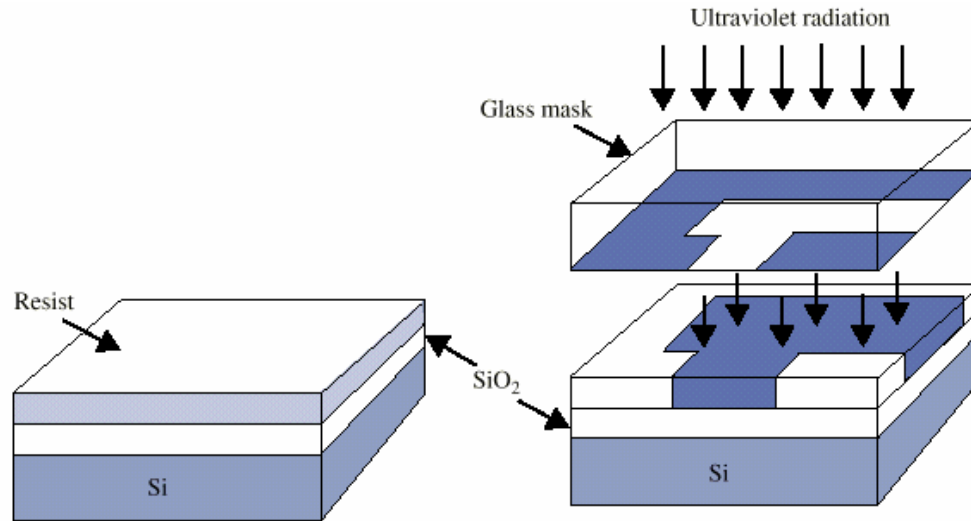
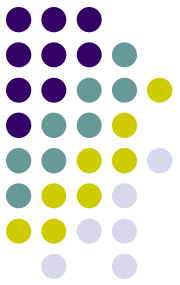


- Reactor Fabrication
- Reactor Characterization
- Reaction Results

Microreactor Device



Fabrication: Photolithography



(a) Coat with photoresist

(b) Expose photoresist
(positive, bonds broken)

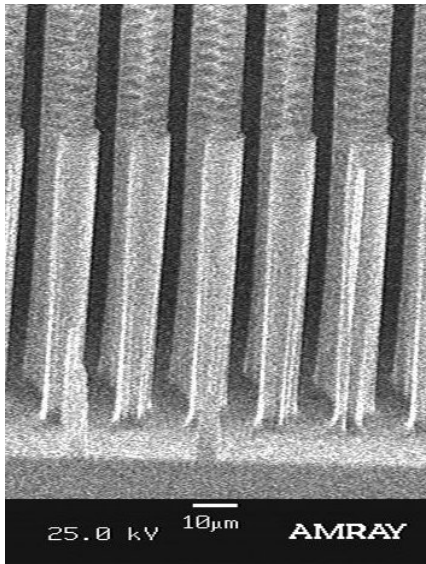


(c) Remove *exposed* resist

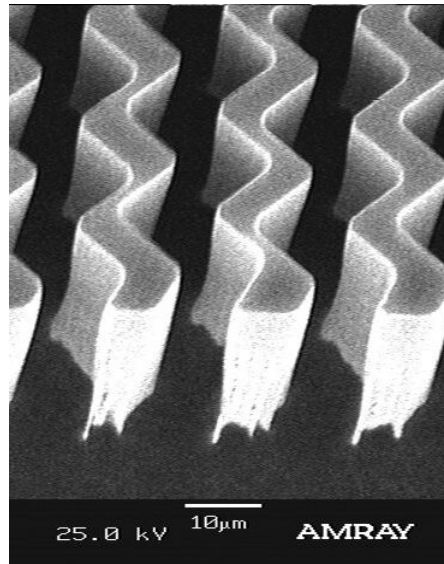
(Shackelford)



Fabrication: Silicon Etching

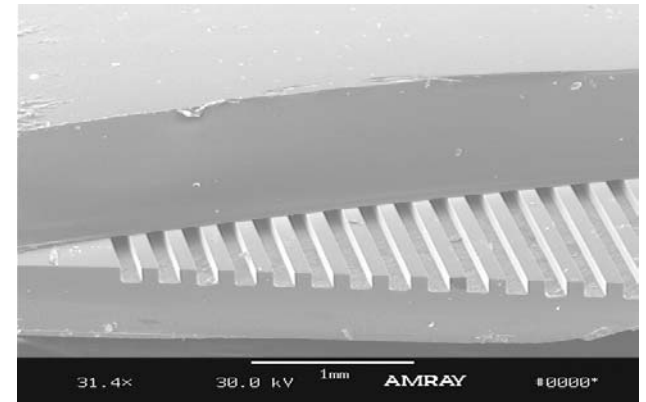
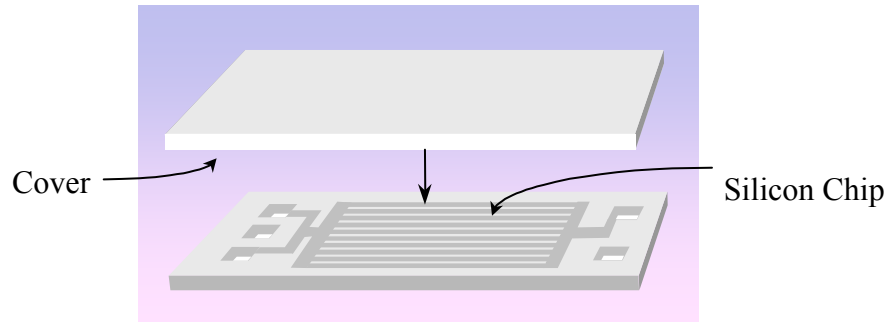


Structured Catalyst Support
in Reaction Zone

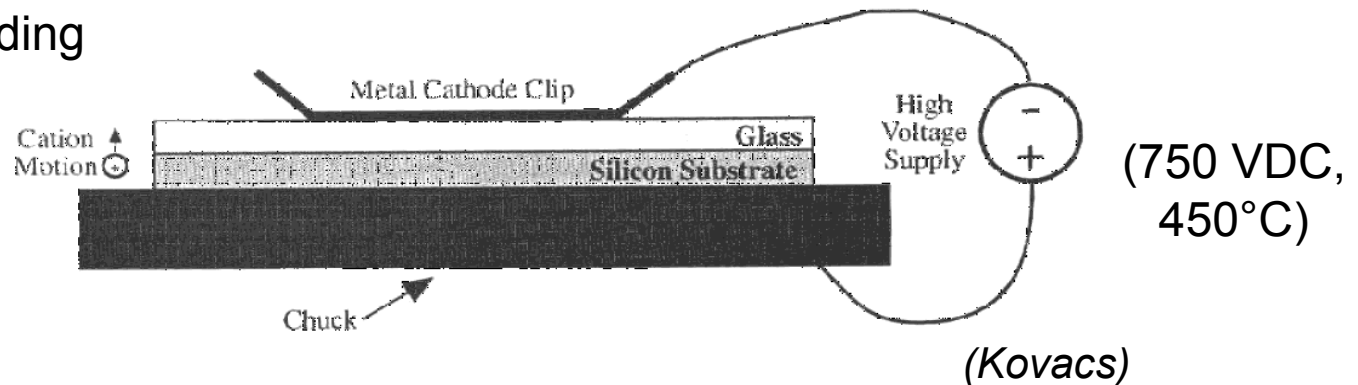


Alcatel Deep Reactive
Ion Etch System

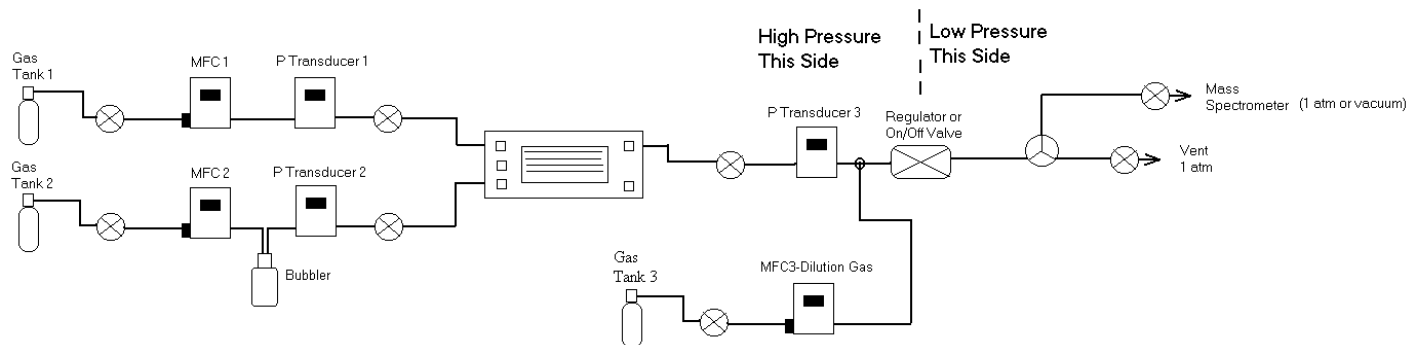
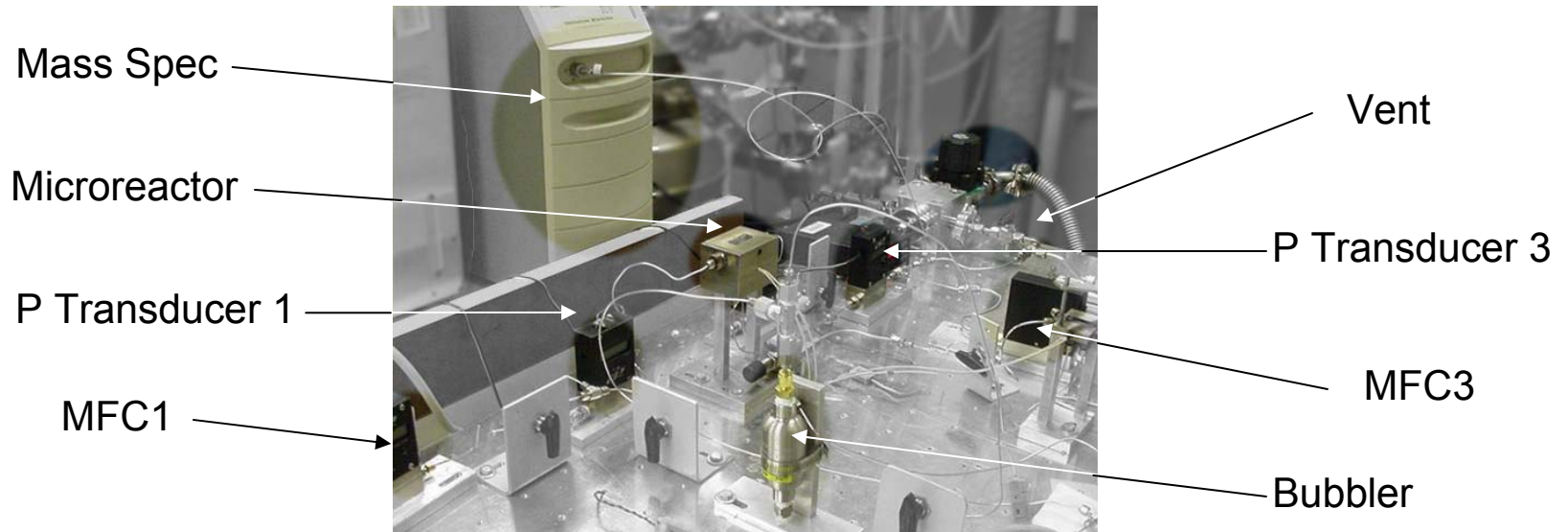
Pyrex-to-Silicon Bonding

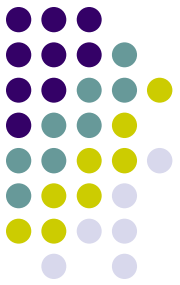


Anodic Bonding



Characterization Experiment Setup

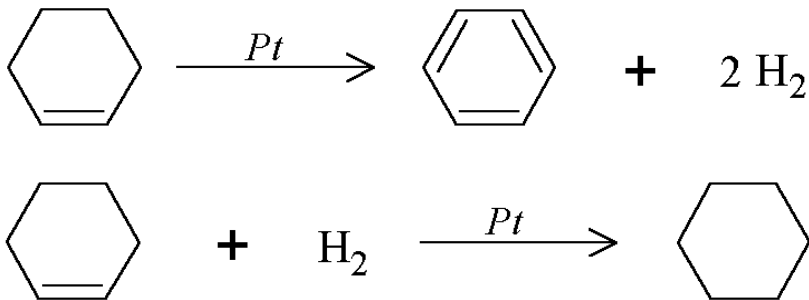
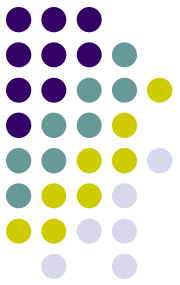




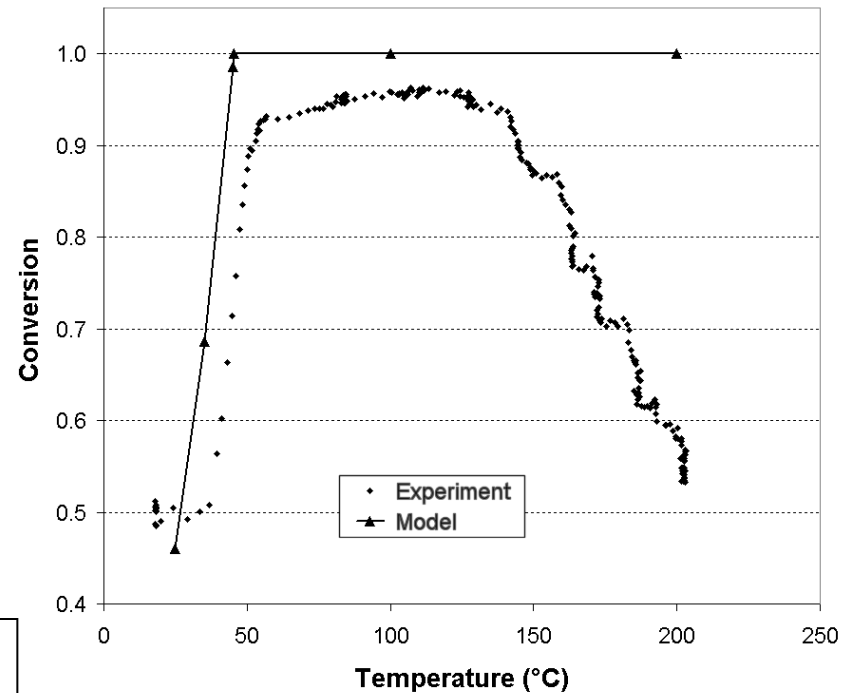
Examples: Reactions

- Hydrocarbon hydrogenation/dehydrogenation
 - Cyclohexene hydrogenation/dehydrogenation
 - Benzene hydrogenation
- Hydrogen + oxygen in explosive regime
- Syngas conversion
- Catalytic combustion

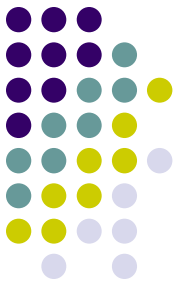
Cyclohexene Hydrogenation/Dehydrogenation



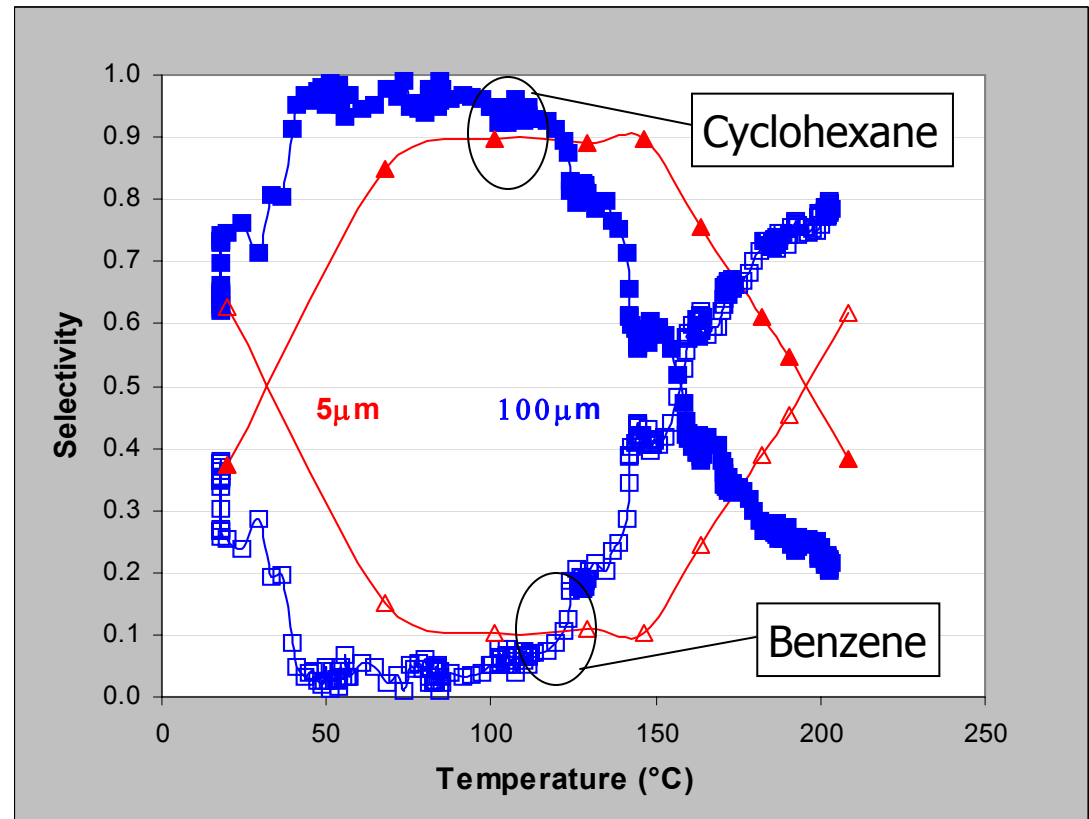
Models for hydrotreating and reforming reactions



Effect of Temperature on Selectivity



- Room temperature activity for both products
- Hydrogenation favored T_{room} to 150°C
- Dehydrogenation favored above 150°C
- Time and temperature dependent deactivation
- 5 μm reactor more tolerant of T and t



Implementation of Microchemical Systems

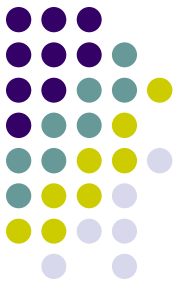


Technology

- Arrays for parallel characterization (R&D)
- On-site, on-demand production
- Special environments
- New factory paradigm

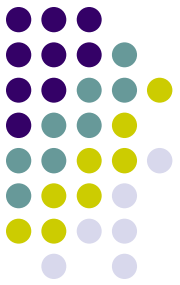
Application

- Catalyst discovery; process development
- Toxics
- Fuel processing (fuel cells)
- Space; offshore platforms
- Highly selective synthesis (pharma, fine chemicals)
- DOE Vision 2020-30% reduction in waste, pollution



Conclusions

- Microreactors possess special properties due to their **small dimensions** ($< 500 \mu\text{m}$).
- Various choices of **robust materials** are available suitable for a variety of applications (metal, ceramic, silicon, polymer).
- **Silicon** microreactor example illustrates reactor fabrication, operation, and characterization.
- ***Model hydrocarbon catalytic hydrogenation and dehydrogenation*** reactions illustrate ability to take relevant reaction engineering data safely and with low consumption.
- Microreaction technology will find a number of **niches** in analytical and process chemistry in the new millenium.



Acknowledgements

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